

# Natural Hydrogen in the northern Semail Ophiolite: A case study in the Emirate of Ras Al Khaimah, UAE

Gabriel Pasquet<sup>a,\*</sup>, Keanu Loiseau<sup>a</sup>, Mohamed Diatta<sup>a</sup>, Giacomo Firpo<sup>b</sup>, Paul Swire<sup>b</sup>, Andrew Amey<sup>b</sup>, Thibaut Burckhart<sup>b</sup>, Isabelle Moretti<sup>a</sup>

<sup>a</sup> Laboratoire des Fluides complexes et de leurs Réservoirs (LFCR), UMR 5150, IPRA, Université de Pau et des Pays de l'Adour, Avenue de l'Université, 64000 Pau, France

<sup>b</sup> RAK Gas, Ras Al Khaimah, United Arab Emirates

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## ABSTRACT

The Semail Ophiolite provides a rare geological snapshot of oceanic lithosphere rocks at the surface. Extensive studies in the southern region, particularly within the Sultanate of Oman, have highlighted active serpentinization processes and the associated natural hydrogen (H<sub>2</sub>) production. However, the northern region of the ophiolite remains underexplored. This study aims to fill this knowledge gap by investigating the potential natural H<sub>2</sub> resources in the Emirate of Ras Al Khaimah (RAK), in the United Arab Emirates (UAE). Comprehensive surface gas measurements allowed us to construct an H<sub>2</sub> distribution map which reveals H<sub>2</sub> presence and, as usual, show that the faults are preferential migration pathways. We also examine the correlation between gas emissions and lithological variations from the metamorphic sole to various mantle peridotites and crustal upper gabbros. Findings indicate that structural windows exposing the metamorphic sole are key areas for locating higher H<sub>2</sub> concentration measurements. Additionally, prominent structural lineaments between the layered and upper gabbros formations represent potential migration pathways for H<sub>2</sub> from the underlying peridotites. New alkaline blue pools (pH~9 and ongoing carbonation) have been identified; the gases sampled in these bubbling pools are very diversified. As in Oman, H<sub>2</sub>-rich gas, up to 88% after air correction, have been found but also N<sub>2</sub> and CH<sub>4</sub>-rich ones. A seasonal variation in the H<sub>2</sub>/N<sub>2</sub> ratio was observed at one site, suggesting significant fluctuations within the underground water table, which is necessary for H<sub>2</sub> generation. These new measurements highlight the northern Semail Ophiolite's potential for natural H<sub>2</sub> exploration and production (E&P) in the RAK Emirate.

## 1. Introduction

The need for a more decarbonized energy mix has been widely recognized by many countries, and many stakeholders - governments, industries, research institutions and individuals - are actively working towards this goal. In this context, CO<sub>2</sub> capture and storage are seen as essential, as some industrial processes remain dependent on CO<sub>2</sub> emissions. The growing use of hydrogen (H<sub>2</sub>) is another promising solution. H<sub>2</sub> is both an essential raw material in the chemical industry and a fuel capable of decarbonizing specific sectors, such as transport and industrial processes. However, the H<sub>2</sub> currently used in industry comes mainly from fossil fuels, such as methane and coal. The development of decarbonated H<sub>2</sub> alternatives is therefore a critical issue. While certain technologies, such as electrolysis, have reached a relatively advanced stage, their cost remains prohibitive compared to conventional methods of H<sub>2</sub> production, such as steam methane reforming. As a result, a

number of alternative methods for producing decarbonized H<sub>2</sub> are being investigated. In many of these approaches, H<sub>2</sub> continues to be synthesized, for example by methane pyrolysis. However, there is a growing interest in exploring natural H<sub>2</sub> as a primary energy source, as it holds the potential to deliver a more environmentally sustainable energy supply [1,2].

### 1.1. Ras Al Khaimah legislation

The Ras Al Khaimah Petroleum Authority (RAKPA) has engaged various external parties with regards to regulating natural H<sub>2</sub>, the conclusion from this work is that existing oil and gas Exploration and Production Sharing Agreements (EPSA's) are best modified for this purpose and this re purposing is currently being undertaken. Concurrently a series of engagements with the geoscience department of the UAE Ministry of Energy and Infrastructure (MOEI) has led to an

\* Corresponding author.

E-mail address: [gabriel.pasquet@univ-pau.fr](mailto:gabriel.pasquet@univ-pau.fr) (G. Pasquet).

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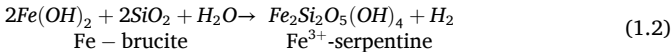
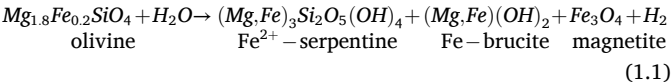
agreement and a Memorandum of Understanding (MOU) has been signed to allow data and information sharing for H<sub>2</sub> and also for Carbon Capture including Mineralization.

An initial third-party evaluation highlighted that significant sub-surface risks remain regarding natural H<sub>2</sub>, necessitating further work to identify and mitigate these risks. RAKPA and RAK Gas, the national oil and gas company, have adopted a two-pronged approach. One of these, the top-down approach, is discussed in this paper. The other utilizing Play Based Exploration (PBE) methodology for a bottom up approach has also been undertaken. Surveys such as Magnetotelluric (MT) are also being acquired to aid in determining thickness variations in the ophiolite across the acreage and to help with prospectivity definition. Initially some shallow wells (to 600 m) are also being planned for profiling the ophiolite and also to acquire fresh unweathered samples. Licensing to interested parties and license monitoring is the natural progression into 2025 and beyond. One party is already committing to exploration through signed agreement initially and other interested parties are also being sought. The RAK road map for decarbonation is shown in Fig. 1.

1.2. Natural H<sub>2</sub>

Natural H<sub>2</sub> has been identified in numerous geological settings and through a wide range of fluid-rock interactions. These various natural H<sub>2</sub> generation contexts have been classified into four groups [3]: (H<sub>2</sub>GR1) oceanic lithospheres, notably including ophiolites and oceanic ridges [4–7]; (H<sub>2</sub>GR2) cratonic and Proterozoic provinces, including Banded Iron Formations (BIFs) [8–11]; (H<sub>2</sub>GR3) uranium-rich provinces [12, 13]; and (H<sub>2</sub>GR4) organic-rich sedimentary basins [14,15].

The H<sub>2</sub>GR1 group is centered on redox reactions between ferromagnesian rocks, such as peridotites, and water. The reduction of water through the oxidation of ferrous iron in olivine generates H<sub>2</sub> (e.g. equation (1.1) [16]). The optimal reaction temperature ranges between 250 °C and 300 °C [17,18]. At low temperatures (<100 °C), the Fe-rich brucite previously produced can be oxidised to form Fe(III)-serpentine (e.g. equation (1.2) [19]) and also minerals like magnetite and hydro-andradite [20], enabling the production of H<sub>2</sub>.



2. The Semail Ophiolite

2.1. Geological settings

The Semail Ophiolite is a region of primary interest for understanding serpentinization processes (H<sub>2</sub>GR1), due to the quality of its outcrops. Various studies, notably through the ICDP Oman Drilling Project [21], have demonstrated the presence of H<sub>2</sub> in bubbling springs

and in the soil of the southernmost massifs around the Jabal Akhdar, including the massifs of Haylayn, Bahla, and Wadi Tayin [22–26]. Our study focusses specifically in the northern part of the Semail Ophiolite, in the Khawr Fakkan and Aswad massifs (Fig. 2).

The Semail Ophiolite is located in the southeast of the Arabian Peninsula, along the northeastern margin of the Sultanate of Oman and the United Arab Emirates, separated from Iran by the Gulf of Oman. This ophiolite nappe extends over 500 km in length and reaches up to 100 km in width in certain areas. It forms part of a mountain range associated with the Alpine orogenic belt [27].

This mountain range is divided into three tectono-stratigraphic units: autochthonous (and para-autochthonous) units, allochthonous units, and post-orogenic units [28]. The autochthonous units consist of a Neoproterozoic crystalline basement, a Proterozoic-Paleozoic sedimentary basement, and a carbonate sedimentary cover that developed from the Middle Permian to the end of the Cretaceous. The autochthonous massifs include the Jabal Akhdar and Saih Hatat (Fig. 2). The allochthonous units comprise the Hawasina sedimentary nappes and the Semail Ophiolite, which were thrust onto the Arabian platform. The Hawasina complex includes the volcano-sedimentary Dibba and Hatta zones. Finally, the post-orogenic units consist of a Cenozoic sedimentary cover that unconformably overlies the older units.

During the Permian, an extensional regime led to continental rifting and the development of intercontinental basins, such as the Hawasina Basin, which represent the early stages of the Neo-Tethys Ocean. This rifting event is linked to the Pan-African orogeny. In the Lower Cretaceous, with the onset of the Alpine orogeny, the tectonic regime shifted to compression, resulting from convergent forces between the Eurasian and Arabian plates [29] this marked the initial phase of the Neo-Tethys Ocean’s closure. Two significant volcanic episodes were recorded during the Cenomanian: V1 is related to the accretion of oceanic crust at a mid-ocean ridge while V2 is associated with the closure of the Neo-Tethys Ocean [30,31]. Several models were proposed to explain the formation of the Semail Ophiolite. One model suggests intra-oceanic thrusting [29], while others propose a fore-arc or rear-arc supra-subduction zone setting [32]. During the Turonian-Campanian period, oceanic detachment continued, culminating in the thrusting of oceanic sequences. Between 85 and 82 Ma, these oceanic units were accreted onto the passive continental margin of Oman [33]. The Maastrichtian marked the end of the obduction process, leading to the final emplacement of the Semail Ophiolite and subsequent isostatic rebalancing of the region. Following the obduction, the Omani continental margin experienced a series of transgression-regression cycles from the Maastrichtian to the Oligocene. Between the Oligocene and the early Miocene, during the Alpine orogenic phase, the Arabian plate continued its northeastward motion in relation to the Red Sea and Gulf of Aden opening [34–36]. These tectonic movements caused the uplift and subsequent erosion of the mountain ranges in Oman and the UAE, leading to the ophiolitic nappe being fragmented into 12 distinct massifs.

The Semail Ophiolite preserves the composition and structure of

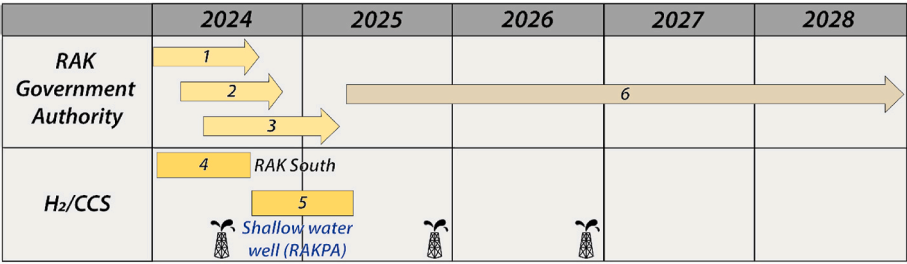


Fig. 1. RAK road map for decarbonation. CCS for Carbon Capture and Storage. The well icon corresponds to the planning of shallow wells for fresh sample acquisition. (1) Regulation drafting. (2) Liaison across government departments. (3) Implementation. (4) H<sub>2</sub> studies. (5) Magnetotelluric acquisitions. (6) Licensing and monitoring.

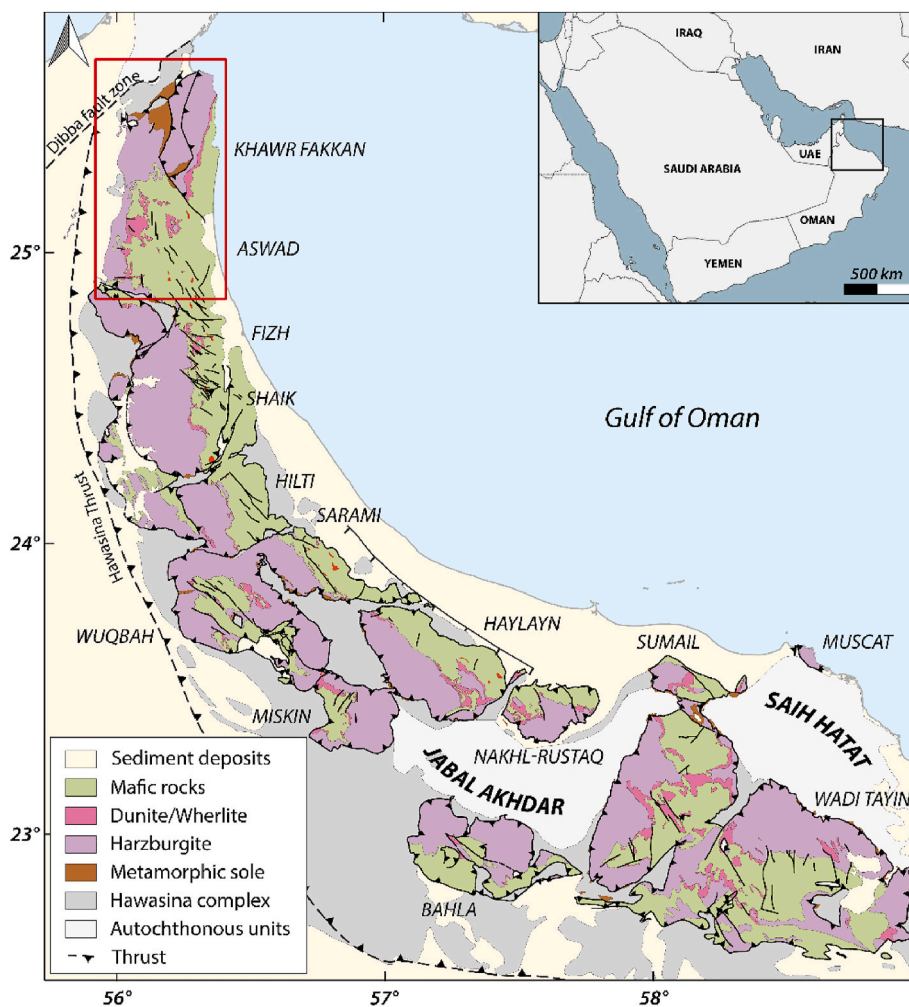


Fig. 2. Lithological units of the Semail Ophiolite, with the study area outlined in red (modified after Nicolas et al. [38]). (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

Tethyan oceanic lithosphere, corresponding to a fast-spreading ridge environment [37,38]. It is primarily composed of pillow basalts, sheeted dikes, gabbros, and upper mantle peridotites, arranged in a stratigraphic sequence from top to bottom [29,39,40]. The upper mantle section, consisting of harzburgites and dunites, exhibits a thickness ranging from 12 to 15 km.

The lateral movement of the ophiolite over the underlying units has resulted in the formation of a metamorphic sole, which developed under pressure-temperature (P-T) conditions reaching 770–900°C and 11–15 kbar. This metamorphic basement, identified in the Bani Hamid and Masafi tectonic windows (UAE), is characterized by a vertical succession of granulite, amphibolite, and schist from top to bottom [41].

## 2.2. Hydrothermalism in the ophiolite

Throughout its evolution, the oceanic lithosphere experienced several episodes of alteration due to various fluid circulation processes. These included oceanic hydrothermal circulation, fluid movement during intra-oceanic detachment, and subsequent surface and meteoric fluid circulation after the emplacement of the ophiolite [42]. These fluid flows facilitated the serpentinization and carbonation of peridotites [17]. It is challenging to estimate the inherited rate of serpentinization during the initial stages of oceanic lithosphere formation at the ridge and off-axis, as well as that acquired during obduction and current onshore hydrothermal processes. However, after a petrographic study in which the different vein generations are distinguished, oxygen isotope

analyses on serpentine or calcite can provide formation temperature data [43]. These data may be correlated with the various geodynamic stages [44]. Current serpentinization levels, in the Wadi Tayin massif, are estimated to average approximately 62% in harzburgites and 81% in dunites [39]. However, average serpentinization rates may not accurately reflect local variability and may differ substantially at all scale across a massif and, consequently, across other massifs as well [45].

Ongoing low-temperature (23°C–60 °C [20]) water-rock reactions, such as serpentinization and carbonation, of peridotites are evidenced in the Oman Drilling Project (DP) wells [46] and at alkaline spring sites, called blue pools, through travertine precipitation and carbonate veins within mantellic rocks [17,47]. In these springs, free gas is observed, with compositions divided into three main types reaching up to 93.8 vol % H<sub>2</sub>, 92.0 vol% N<sub>2</sub> and 50.5 vol% CH<sub>4</sub> [22,24,25]. Often associated with active serpentinization within the ophiolite, H<sub>2</sub> is also detected in the soil of sub-ophiolitic units in the Jabal Akhdar and Hawasina regions [26].

## 3. Materials and methods

In an ophiolitic context, the main process to generate H<sub>2</sub> is the oxidation of Fe-rich minerals associated to the reduction of water [48–50]. To confirm the presence of deep H<sub>2</sub> generation, or leakage from accumulations, abnormal H<sub>2</sub> content in the soil or spring were measured and sampled [3,51].

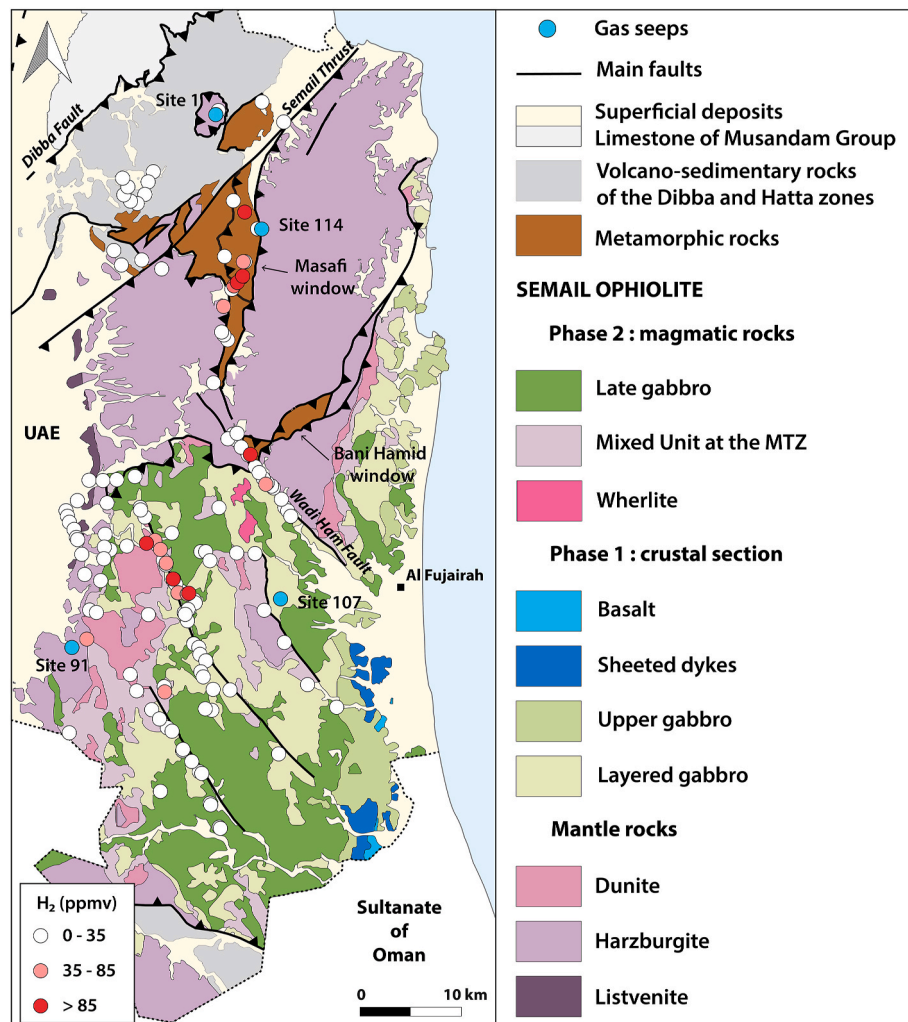
### 3.1. Soil gas analysis

For soil measurements, we used a field gas analyzer called BIOGAS 5000. Our version measures six gases: CH<sub>4</sub> (%), CO<sub>2</sub> (%), O<sub>2</sub> (%), H<sub>2</sub> (ppmv), CO (ppmv), and H<sub>2</sub>S (ppmv), as well as the balance (%), which corresponds to non-analyzed gases and is typically interpreted as being close to the N<sub>2</sub> content. The analyzer measures CH<sub>4</sub> and CO<sub>2</sub> using a dual-wavelength infrared sensor, while O<sub>2</sub>, CO, H<sub>2</sub>, and H<sub>2</sub>S are detected using internal electrochemical cells. The measurement uncertainty for the values mentioned in the results is  $\pm 0.5\%$  for CH<sub>4</sub> and CO<sub>2</sub>,  $\pm 1.0\%$  for O<sub>2</sub>, and  $\pm 2.5\%$ ,  $\pm 2.0\%$ , and  $\pm 2.0\%$  for H<sub>2</sub>, CO, and H<sub>2</sub>S, respectively. The depth of the measurement is classically between 20 and 80 cm (length of the drilling bit), when rocks aren't directly outcropping in these areas where few soils have been developed. Recent measurements have shown that H<sub>2</sub> values between 1 and 5 m are not constant but remain within the same order of magnitude [52]. This behavior is similar to that of CO<sub>2</sub>, as observed by Klusman [53], despite an increase of few percent in concentration between 30 and 100 cm. Shallow measurements consistently yield a high air content and therefore a lower percentage of other gases. Sporadic emanations are measured, then a single value can't be considered as representative [3,54]. Accordingly, a minimum of three measurements was conducted at each location, with a spacing of a few tens of centimeters, but only the maximum values are plotted (Fig. 3 and Table A.1). The resulting mapping can therefore be interpreted in terms of structural patterns related to the H<sub>2</sub> migration

pathways of a given area, rather than a purely quantitative indicator.

### 3.2. Bubbling spring analysis

We located four bubbling seeps in a water spring (Fig. 3), allowing us to obtain quantitative gas concentration measurements. By analogy with alkaline to hyperalkaline springs with H<sub>2</sub>-rich seeps, as described in the Oman case study [22,23,26], we draw comparisons to our findings. The gas was sampled by collecting bubbles in a septum-sealed funnel and injected into 10 mL vials using a gas tight syringe. The gas-tight syringe prevents gas exchange with the atmosphere before injection. These 10 mL vials allow us to determine the gas composition by gas chromatography (GC), and coupled with this GC, a mass spectrometer (GC-MS) enables us to analyze the isotopic composition of the gases for hydrogen (H), carbon (C), and nitrogen (N). The H<sub>2</sub>, O<sub>2</sub>+Ar, N<sub>2</sub>, CO<sub>2</sub>, CH<sub>4</sub> and hydrocarbons are analyzed on an Agilent 6890 N/7890A/7890 B GC using three detectors. A TCD detector for H<sub>2</sub>, O<sub>2</sub>+Ar, N<sub>2</sub> and CH<sub>4</sub>, another TCD detector for CO<sub>2</sub> and a FID detector for hydrocarbons. From the results of all three detectors, one complete composition is calculated. Detector responses are calibrated several times a day using various reference standards. At least three vials were collected per bubbling site: two for major gas compositions (including one duplicate) and one for isotopic compositions. For the major gas composition, standard deviations are below 2% for concentrations above 1 mol% and below 10% for concentrations below 1 mol%. For isotopic compositions, three



**Fig. 3.** H<sub>2</sub> soil measurements displayed on the geological map of the Semail Ophiolite centered on the UAE (see location Fig. 2), as well as gas sampling sites in blue (geological contours after [56]). (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

measurements are performed on the same vials to assess measurement repeatability. The reported results represent the average of these three measurements, with standard deviations provided in Table 3. Furthermore, measurements of pH and water temperature were taken directly on site using a portable analyzer.

## 4. Results

### 4.1. Soil gas

In combining different missions, 488 gas measurements were carried out at 145 sites in the study area. Not surprisingly, Fig. 3 highlights the role of the faults in the fluid migration, and a trend emerges with the highest values on the large NW-SE lineaments in Ras Al Khaimah (RAK) South.

The highest values, in addition to being on major lineaments, are at the contact between the Layered Gabbro (GbL) and Mixed Unit (Mu) formations, with 121 ppmv, in RAK South. This transition between Mu and GbL is the crust-mantle transition.

To the north and northeast, the highest values are found in the sub-ophiolitic metamorphic windows. In the Masafi Shale Formation (MSF4), with 88 ppmv, for the Bani Hamid window; and in the Tayyibah Metavolcanic formation (TMv), with 171 ppmv, for the Masafi window.

Using a quantile-quantile graphical representation [55] of soil H<sub>2</sub> data (Fig. 4), two inflection points are observed, the first at 35 ppmv H<sub>2</sub> and the second at 85 ppmv H<sub>2</sub>. From 0 to 35 ppmv, the values correspond to the soil gas atmosphere background in the area. From 35 to 85 ppmv, the values are more frequent and correspond to areas near potential H<sub>2</sub> emission sources close to the main faults, while values above 85 ppmv likely represent active H<sub>2</sub> emission zones above main faults, according to the geological map. It should be noted that outside main lineaments values are often below the 35 ppmv anomaly (Fig. 3). Values above the second anomaly identified at 85 ppmv (Fig. 4) are located along the main southern lineament between GbL and Mu, as well as in the tectonic windows of Bani Hamid and Masafi.

Fig. 3 also shows the 4 sites where free-gas were sampled. Site 1 corresponds to a bubbling blue-pools on a klippe of harzburgite north of the Semail Thrust. Site 91 corresponds to a wadi on the western front of the ophiolite, in the harzburgites of RAK South. Site 107, is located in a wadi on the GbL Fm. east of the ophiolite. Site 114, is located in a wadi

above the harzburgites, near the Masafi metamorphic window.

### 4.2. Bubbling springs

Gas composition and isotopes measurements were carried out on these 4 different sites (Figs. 3 and 5), representing 5 samples because two were taken at site 1. The results are listed in Tables 1–3 below. The gases, after air correction [57,58], appear to be blends. Results show that 3 major end-members stand out, the H<sub>2</sub>-rich type with site 114, the CH<sub>4</sub>-rich type with site 91 and the N<sub>2</sub>-rich type with sites 1. It should be noted that site 107 shows an abnormally high oxygen concentration. Also, all sites are poor in CO<sub>2</sub>.

Table 2 and Fig. 5 show temperatures (measured 5 cm below the water surface) ranging from 25.5 to 29.3 °C and pH levels between 8.09 and 9.31 in these blue pools and wadis where gas samples were collected. Fig. 5a illustrates the alkaline nature of Site 1 and 1', where substantial carbonate precipitation is observed, appearing as travertine or carbonate veins within the serpentinized massif. These carbonate precipitations were also observed in the wadi of Site 114, but were absent in the wadi Site 91 and 107.

The isotopic values in Table 3 show that the most H<sub>2</sub>-rich sites with 44.6% and 87.8% (see Table 2), respectively on 1' and 114, have very homogeneous  $\delta^{15}\text{N}$ ,  $\delta^{13}\text{C}$  and  $\delta\text{D}$  isotopic values. The equilibrium temperatures associated with these sites on the CH<sub>4</sub>-H<sub>2</sub> pair [59] are 66.6 °C and 44.9 °C for 1' and 114 respectively. Site 91, rich in CH<sub>4</sub>, shows relatively similar isotopic values for  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\delta\text{D}_{\text{CH}_4}$ , but a  $\delta^{13}\text{C}_{\text{CH}_4}$  reaching values of −61.5‰. Giving an equilibrium temperature in CO<sub>2</sub> and CH<sub>4</sub> [60] around 108 °C.

## 5. Discussion

### 5.1. Soil gas

As shown by the soil measurements in the field, the high H<sub>2</sub> values (>85 ppmv of H<sub>2</sub>, Fig. 3) are associated with seepage along large faults bringing the GbL and Mu units into contact at surface and connecting at depth in the mantle peridotites. Soil measurements have also shown that the metamorphic basement, observed in the Masafi and Bani Hamid tectonic windows, on which the ophiolite has moved on, can play the role of H<sub>2</sub>-generating rock. Values are higher in the metamorphic sole as observed in the Jabal Akhdar [26]. The fact that no H<sub>2</sub> is found in the non-fractured outcropping mantle and gabbro units (<35 ppmv), as also observed in southern massifs [26], may suggest that the gabbro units act as a cap for the H<sub>2</sub> system. Indeed, these gabbros are low in porosity and permeability compared to the mantle section below. The GbL from the CM1A well in the Wadi Tayin massif, drilled during the Oman DP campaigns, indicate a transport porosity of 0.12% and a calculated permeability from the core resistivity of  $5.39 \times 10^{-21} \text{ m}^2$  [61]. The dolerites sealing the H<sub>2</sub> reservoir at Bourakebougou have a total porosity of 0.39% [62]. However, the permeability of the gabbros may vary at the massif scale depending on fracture density. The large draining faults facilitate the migration of H<sub>2</sub> (Fig. 3) generated in the underlying volcanic, sediment-rich mantle or metamorphic basement. In the different context of stimulated H<sub>2</sub> by injecting water in ophiolites, the gabbro could be a lateral seal [63].

### 5.2. Bubbling springs and dry gas seepages

#### 5.2.1. Gas origins

Site 114, rich in H<sub>2</sub>, is totally consistent with gas measurements made in the southern part of the ophiolite in Oman and described in the literature as shown in Fig. 6. The N<sub>2</sub> pole of site 1 is generally observed in sediment-hosted formations in Jabal Akhdar (Fig. 6). Site 1', after air correction, is close to the N<sub>2</sub>-H<sub>2</sub>-rich group, commonly measured in the ophiolites of New Caledonia, Philippines and Oman [22,25]. CH<sub>4</sub>-rich sites are uncommon in Semail and have more often been described in

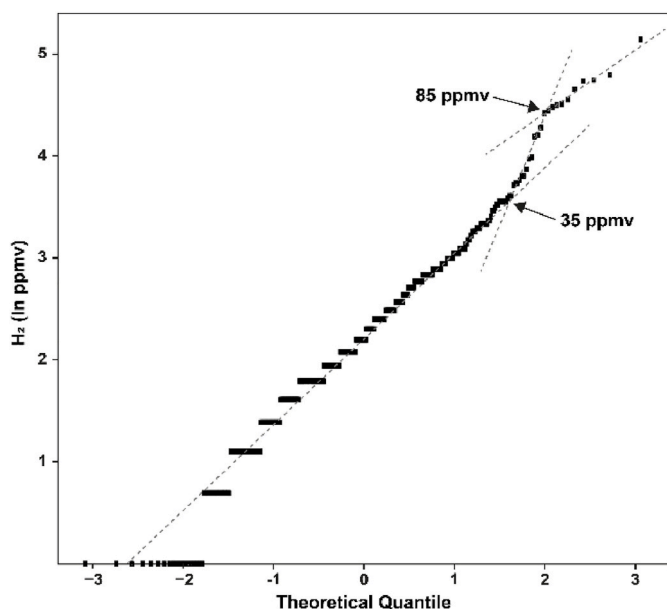
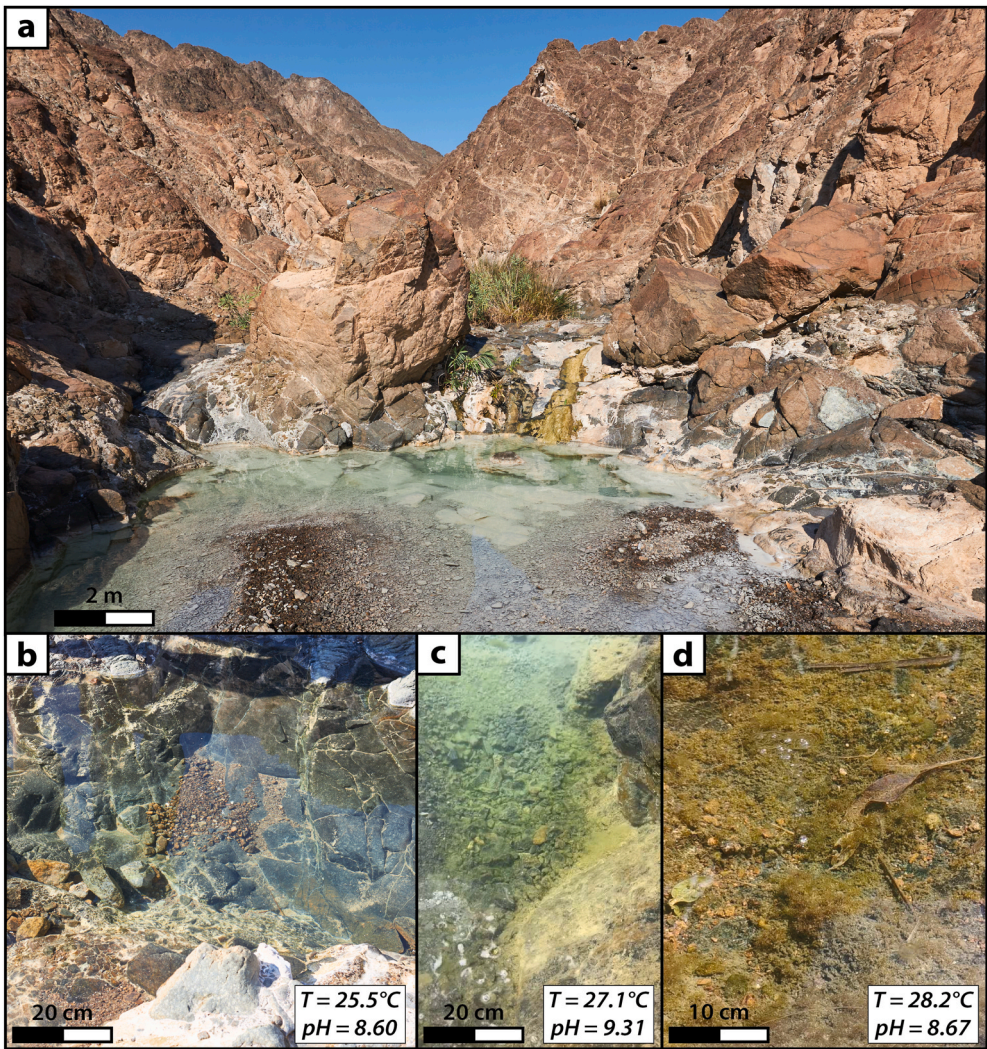


Fig. 4. Diagram showing the quantile-quantile plot calculated [55] for H<sub>2</sub> with two anomaly threshold values corresponding at 35 ppmv and 85 ppmv.



**Fig. 5.** Photographs of the various gas sampling sites. (a) Alkaline springs on the Harzburgite klippe downstream from (b) sites 1 and (c) 1'. (d) Carbonated spring rich in H<sub>2</sub> at site 114.

**Table 1**  
Gas composition, in mol%, sampled at the various bubbling sites during the two surveys. Site 1 corresponds to bubbles taken during the first field acquisition, and site 1' from the same area during the second field acquisition. “bdl” stands for “below detection limit”, corresponding to 0.04 mol% for H<sub>2</sub> and H<sub>2</sub>S, 0.01 mol% for CO<sub>2</sub>, O<sub>2</sub>, and N<sub>2</sub>, and 10<sup>−4</sup> mmol% for hydrocarbons.

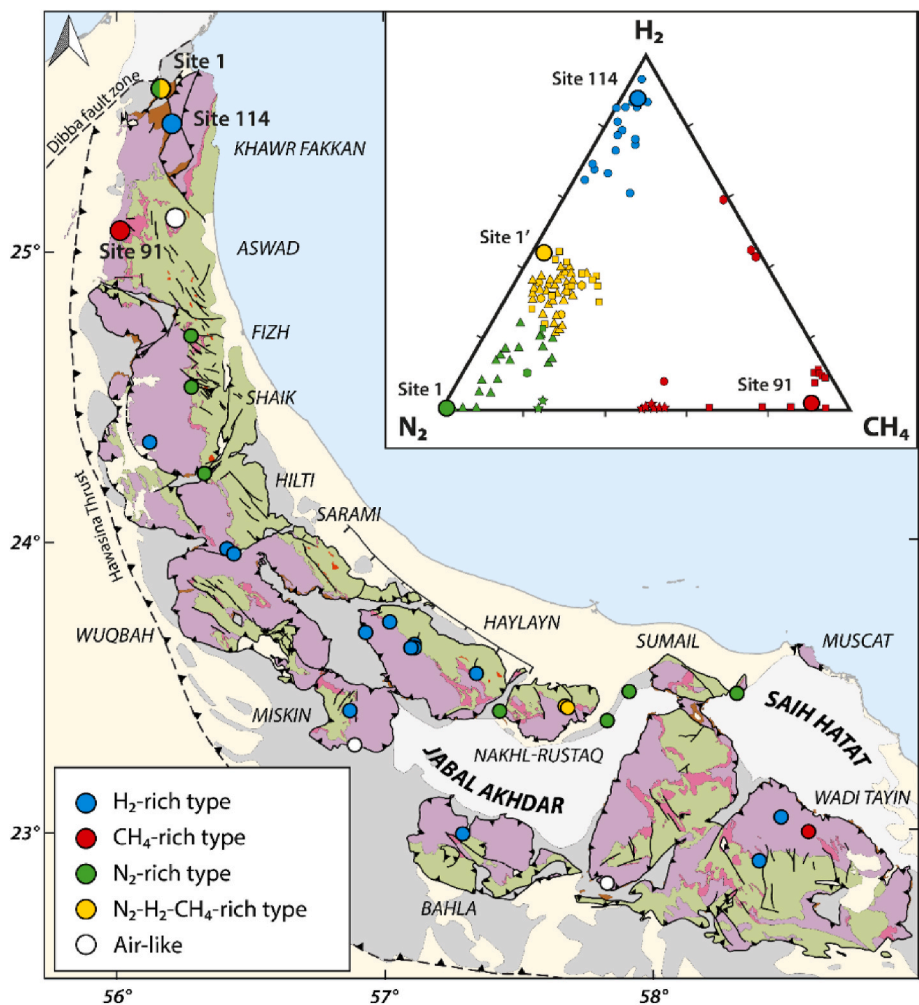
Sites	Latitude	Longitude	H <sub>2</sub>	CH <sub>4</sub>	N <sub>2</sub>	O <sub>2</sub>	CO <sub>2</sub>	Ar	H <sub>2</sub> S	C <sub>2</sub> H <sub>6</sub>	C <sub>3</sub> H <sub>8</sub>
			mol%	mol%	mol%	mol%	mol%	mol%	mol%	mmol%	mmol%
1	25.54399	56.155203	bdl	0.06	95.4	3.5	0.04	1.00	bdl	0.50	bdl
1'	25.5429384	56.15553363	16.50	0.22	69.20	13.20	0.01	0.84	bdl	0.50	1.00
91	25.07151935	56.01544456	bdl	78.30	16.20	2.40	2.60	0.46	bdl	0.60	0.20
107	25.114256	56.21195786	bdl	0.35	63.00	35.70	0.10	0.82	bdl	bdl	bdl
114	25.44023688	56.19892178	79.70	3.30	14.7	1.90	0.04	0.26	bdl	0.60	bdl

**Table 2**  
Gas composition after air correction, in mol%, sampled at the various bubbling sites during the two surveys.

Sites	T	pH	H <sub>2</sub>	CH <sub>4</sub>	N <sub>2</sub>	CO <sub>2</sub>	Ar	C <sub>2</sub> H <sub>6</sub>	C <sub>3</sub> H <sub>8</sub>	C <sub>1</sub> /(C <sub>2</sub> +C <sub>3</sub> )
	(°C)		mol%	mol%	mol%	mol%	mol%	mmol%	mmol%	
1	25.5	8.60	bdl	0.07	98.87	0.04	1.01	0.60	bdl	110
1'	27.1	9.31	44.64	0.59	54.08	0.00	0.68	1.35	1.00	253
91	26.3	8.09	bdl	88.47	8.20	2.93	0.40	0.68	0.20	100,761
107	29.3	8.73	bdl	0.00	98.91	0.00	1.09	Bdl	0.10	–
114	28.2	8.67	87.75	3.63	8.39	0.04	0.19	0.66	bdl	5500

**Table 3**  
Isotopic compositions of gases at different sites, for  $\delta^{15}\text{N}$ ,  $\delta^{13}\text{C}$ ,  $\delta\text{D}$  and standard deviations in index of isotopic values, associated with gas equilibrium temperatures calculated for  $T_{\text{CO}_2\text{-CH}_4}$  [60] and for  $T_{\text{CH}_4\text{-H}_2}$  [59]

Sites	$\delta^{15}\text{N}$		$\delta^{13}\text{C}$				$\delta\text{D}$				$T_{\text{CO}_2\text{-CH}_4}$ (°C)	$T_{\text{CH}_4\text{-H}_2}$ (°C)
	% vs atm		% vs PDB				% vs SMOW					
	N <sub>2</sub>		CH <sub>4</sub>		CO <sub>2</sub>		H <sub>2</sub>		CH <sub>4</sub>			
1'	+0.4	(0.0)	−5.9	(0.04)	−17.7	(0.06)	−729	(0.4)	−325	(1.2)		66.6
91	+2.5	(0.03)	−61.5	(0.17)	−15.9	(0.05)	−		−309	(1.0)	108.1	
107	0.0	(0.07)	−49.4	(0.02)	−19.5	(0.13)	−		−304	(0.2)	227.7	
114	+0.4	(0.08)	−10.6	(0.05)	−20.3	(0.09)	−733	(0.2)	−275	(0.3)		44.9



**Fig. 6.** Gas emanation map of the Semail Ophiolite, based on the geological map from Fig. 2 (modified after Nicolas et al. [38]), showing the various bubbling springs measured and the types to which they belong. Ternary diagram illustrating the 3 end-members:  $\text{H}_2$ ,  $\text{N}_2$  and  $\text{CH}_4$  of the different gas springs sampled, in the Semail Ophiolite (big circles are from this study in UAE and small circles from Oman), the Philippines ophiolites (hexagon), the Turkish ophiolites (square), the New Caledonian ophiolite (triangle) and the Italian ophiolites (star) [4,5,22–25,65,66].

Turkish ophiolites as gas seeping in fractures [25].

Fig. 7a shows a  $\text{CH}_4$  origin associated with continental serpentinization for  $\text{H}_2$ -rich sites and a microbial  $\text{CH}_4$  origin for site 91. Sites 1' and 114 are perfectly correlated with measurements made south of the ophiolite in Oman, with low equilibrium temperatures (Table 3 and Fig. 7b). Fig. 7c supports the onshore origin of hydrogen in  $\text{CH}_4$  and  $\text{H}_2$ , and Fig. 7d confirm this abiotic origin of  $\text{CH}_4$  on sites 1' and 114. Regarding  $\text{C}_1/(\text{C}_2+\text{C}_3)$  ratio vs  $\delta^{13}\text{C}_{\text{CH}_4}$ , this microbial origin is associated with a primary microbial reaction, close to methyl fermentation, for site 91 (Fig. 7d).  $\text{O}_2$ -rich sample, site 107, was taken in a water with high algae concentration, and therefore could reflect biosynthesis.

5.2.2. Seasonal variation of the gas content

Site 1, where we went twice, exhibits a temporal variation in gas composition, transitioning from an  $\text{N}_2$ -rich type to an  $\text{N}_2\text{-H}_2$ -rich mixture between November and April. This raises the question of the coexistence of two gas generation zones at the Masafi window, involving a constant crustal emission from the metamorphic sole and an intermittent serpentinization. Indeed, the two reactive zones [25,48] could be: a shallow zone fed by an aquifer with meteoric water circulating through the ultra-basic rocks forming these  $\text{H}_2$ -rich gases; a deeper zone coming from interstitial water, organic matter or ammonium trapped in the metamorphosed sediments of the sub-ophiolitic units. The  $\text{N}_2\text{-H}_2$  mixture would then be a blend of these two sources. Thus, at site 1,

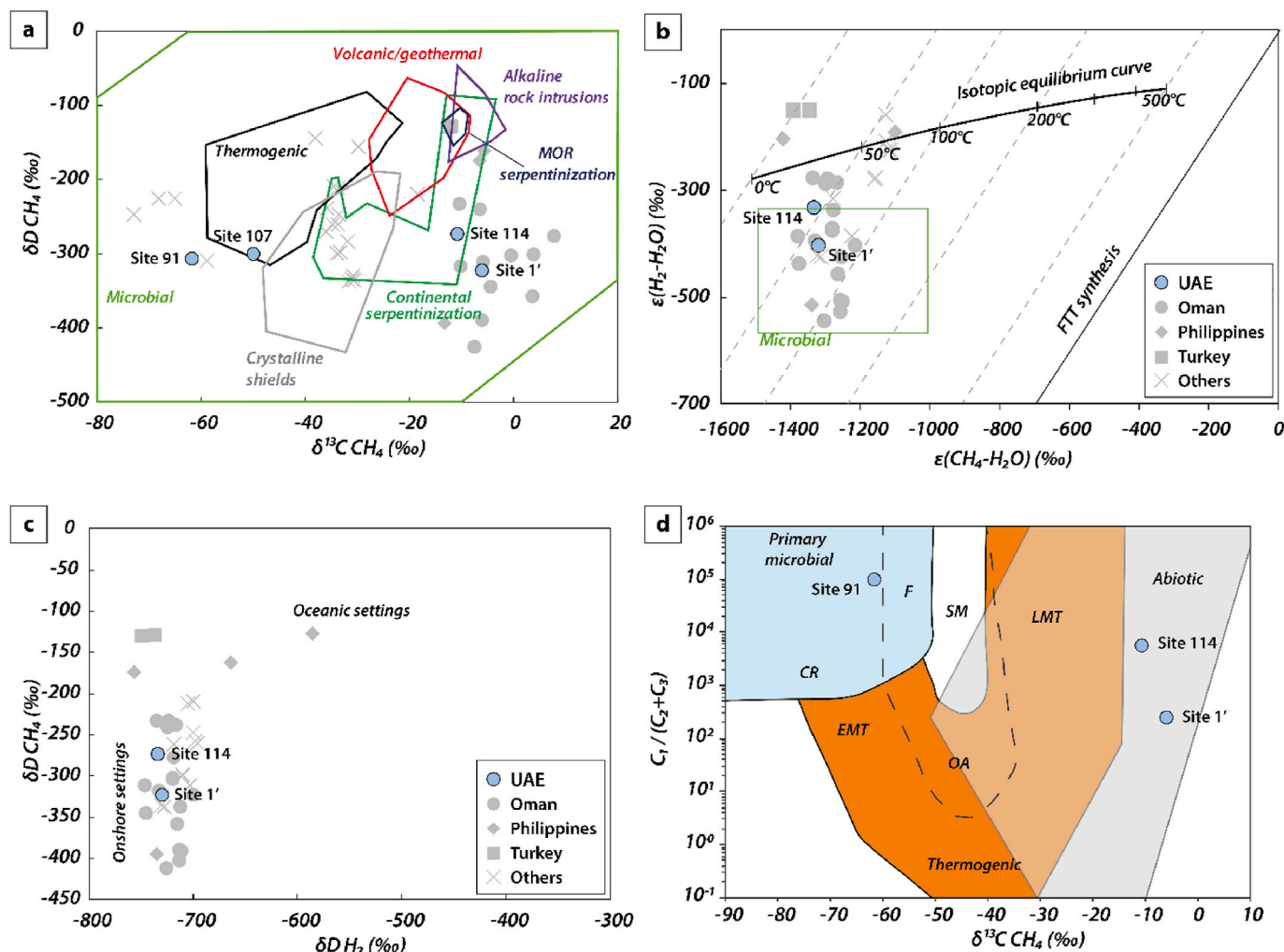


Fig. 7. Isotopic data from Semail Ophiolite sites, UAE. (a)  $\delta D CH_4$  as a function of  $\delta^{13}C CH_4$ , where different boxes are used to represent the origin of  $CH_4$  (modified from Lévy et al. [3], Etiope [67], and Xia et al. [68]). (b)  $\epsilon$  notation of  $H_2-H_2O$  and  $CH_4-H_2O$  pairs, representing whether the system is at thermodynamic equilibrium and determining the  $H_2-CH_4$  equilibrium temperature (modified from Lévy et al. [3]). (c)  $\delta D CH_4$  as a function of  $\delta D H_2$  to compare the onshore or offshore nature of  $\delta D$  (modified from Vacquand et al. [25]). (d) Diagram of  $C_1/(C_2+C_3)$  ratio as a function of  $\delta^{13}C CH_4$ . With the genetic fields: CR =  $CO_2$  reduction, F = methyl-type fermentation, SM = secondary microbial, EMT = early mature thermogenic gas, OA = oil-associated thermogenic gas, and LMT = late mature thermogenic gas (modified after Ketzner et al. [69]).

which varies between the  $N_2$  and  $N_2-H_2$ -rich types, the hypothesis would be the presence of a non-permanent superficial aquifer that causes the gas ratios to vary when it is present or not. This hypothesis can be correlated with the regional climate, as the arid conditions here include a dry season extending from May to October, followed by a rainy season from November to March. Our samples from this site indicate 98.9%  $N_2$  in November, just after the dry season, and 44.6%  $H_2$  with only 54.1%  $N_2$  in April, following the rainy season. We interpret this change as being due to the recharge of an aquifer or, at least, increased water saturation of the rocks.

Based on the temporal variability of gas concentrations and the isotopic  $\delta^{13}C$  and  $\delta D$  data measured at  $H_2$ -rich sites 1' and 114 (Table 3 and Fig. 7), we can infer that the  $H_2$  system is active, currently driven by onshore low-temperature serpentinization (between 45 and 66 °C), as suggested in Al Farfar (Bani Hamid window) [64].

At site 114, serpentinization of the peridotites is dominant, and this  $N_2$ -rich crustal source does not appear to be present or is highly diluted. This suggests the presence of a permanent shallow aquifer in the area.

$\delta^{15}N$  does not allow us to make final conclusions about the origin (atmospheric via an aquifer or crustal) of the  $N_2$ -rich gases in the ophiolites, analysis of noble gases will be more conclusive.

### 5.2.3. Regional variation of the gas content

Several factors may explain the variation in gas content observed throughout the Semail ophiolite (Fig. 6).

The primary lithological composition of peridotites can influence the measured hydrogen content. As shown by Stevens and McKinley [70], fayalite, followed by forsterite, will produce more  $H_2$  during RedOx processes than pyroxene. Thus, for the same fluid, a serpentinized dunite will generate more  $H_2$  than a harzburgite, which itself will produce varying amounts of  $H_2$  depending on its pyroxene content (especially orthopyroxene). Similarly, fluid input into the system plays a major role in this production, as it enables the reduction processes responsible for  $H_2$  formation. As demonstrated by Reed and Palandri [71], the nature of the fluid, whether fresh or saline, impacts both  $H_2$  generation and transport. The accessibility of the fluid to these various Fe-bearing minerals is crucial. A higher connected porosity (and therefore higher permeability) increases the reactive surface area, potentially leading to greater  $H_2$  production.

As shown by the seasonal variability discussed in the previous section, the timing of measurements across different massifs can influence these values, leading to results that fluctuate between  $H_2$  and  $N_2$  end-members.

Finally,  $CH_4$  content is likely more dependent on the environmental

conditions in which these measurements are taken. A setting favorable to microbial activity could lead to higher CH<sub>4</sub> concentrations. Additionally, CH<sub>4</sub> may also form abiotically through the reaction of H<sub>2</sub> with CO or CO<sub>2</sub> via Fischer-Tropsch or Sabatier reactions. The CO and CO<sub>2</sub> involved in these processes can have multiple origins within the ophiolite, including mantle degassing, carbonate dissolution from the numerous pre-ophiolitic carbonate formations (e.g. Hawasina nappes), or oxidation of organic matter present in the pre-ophiolitic sediments.

## 6. Conclusion

These new gas data from the northern zone of the Semail Ophiolite demonstrate once again the suitability of this region for natural H<sub>2</sub> exploration. These data are very encouraging for potential future production, with gas emanation up to 87% of H<sub>2</sub> at surface in alkaline springs. In addition, soil and gas seeps data over time highlight the presence of a more or less continuous flow towards the surface along major faults in the ophiolite and tectonic windows (>85 ppmv of H<sub>2</sub>). Soil gas measurements indicate a near surface background level <35 ppmv of H<sub>2</sub> within the ophiolite and the metamorphic sole. The upper crustal units (gabbros) may act as a seal for gas emissions.

The gas analyses are fully in line with what has been observed in the Sultanate of Oman, with the presence of the 3 major gas end-members (H<sub>2</sub>-rich, N<sub>2</sub>-rich and CH<sub>4</sub>-rich) in the study area. The H<sub>2</sub>-rich gases are interpreted as being associated with low-temperature continental serpentinization, with minima of 45 °C and 66 °C, based on isotopic values. The temporal variation in gas composition from N<sub>2</sub> to N<sub>2</sub>-H<sub>2</sub> at Site 1, within the alkaline spring of the Harzburgite klippe, suggests continuous deep N<sub>2</sub> generation of crustal metamorphic origin and periodic shallow H<sub>2</sub> generation through serpentinization when groundwater circulation occurs. Water supply in this region appears to play a crucial role in the availability of H<sub>2</sub> resources. No source containing free gas has been observed in tectonic windows where the metamorphic sole is exposed to confirm soil measurements. The CH<sub>4</sub>-rich source is interpreted as a dry gas seepage from a microbial origin.

The diversity of H<sub>2</sub>-generating rocks, both mantellic and crustal metamorphic, only increases the field of possibilities for these sites of interest.

## CRediT authorship contribution statement

**Gabriel Pasquet:** Writing – review & editing, Writing – original draft, Visualization, Methodology, Investigation, Formal analysis, Data curation, Conceptualization. **Keanu Loiseau:** Writing – review & editing, Investigation, Formal analysis. **Mohamed Diatta:** Writing – review & editing, Investigation, Formal analysis. **Giacomo Firpo:** Writing – review & editing, Validation, Supervision, Resources, Investigation. **Paul Swire:** Writing – review & editing, Validation, Resources, Investigation. **Andrew Amey:** Validation, Project administration, Funding acquisition. **Thibaut Burckhart:** Writing – review & editing, Validation, Project administration, Funding acquisition. **Isabelle Moretti:** Writing – review & editing, Validation, Supervision, Project administration, Investigation.

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## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.ijhydene.2025.02.259>.

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